THE ABSOLUTE CONFIGURATION OF THE NEW AMINO ACID 2-AMINO-4-METHYL-HEX-5-ENOIC ACID FROM A NEW GUINEA BOLETUS*

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Key Word Index-Boletus; Boletaceae; amino acids; absolute configuration.

Abstract—The absolute configuration of the 2-amino-4-methyl-hex-5-enoic acid isolated from *Boletus* was shown to be 2S, 4S, by an unambiguous synthesis of its dihydro derivative from 2S-(-)-2-methylbutan-1-ol.

We have previously reported the isolation of 2-amino-4methyl-hex-5-enoic acid (1) [1,2] and shown that the asymmetric center at carbon-2 was S but could not establish the configuration at carbon-4 by physicochemical methods. We have now synthesized (2\$,4\$)-2-amino-4-methyl hexanoic acid (2) unambiguously from 2(S)methylbutan-1-ol and acetylamino malonate [3, 4], followed by resolution of the carbon-2 center of the acetylaminomethylhexanoic acid by hog kidney acylase I [5, 6]. The synthetic product was identical in all respects with the dihydro compound (2) obtained from the natural product (1) by catalytic hydrogenation. As catalytic hydrogenation of the $\Delta 5$ double bond at room temperature will not change the chirality at C-4, the absolute configuratio of 1 is established as (2S,4S)-2-amino-4-methylhex-5-enoic acid.

The unambiguous synthesis of sterically pure 2 also establishes the stereochemistry of the naturally occurring amino acid, homoisoleucine, which had been isolated by Fowden and Smith [7] from Aesculus californica. Fowden et al. [8] had already proposed the 2S,4S configuration for homoisoleucine on the basis of a comparison of its solubility, ORD and CD with those of isoleucine.

EXPERIMENTAL

Mps are uncorr. Optical rotations were determined on an ETL-NPL automatic polarimeter.

Hydrogenation of 2-amino-4-methyl-hex-5-enoic acid (1). 0.1 g 1 over Pt required one mole of hydrogen (15.4 ml) and gave (2S, 4S) 2-amino-4-methylhexanoic acid (2) (0.1 g) mp 240-246°; m/e 146, MH⁺ from CI; $\lceil \alpha \rceil_D^{2^2} + 34.3^\circ$ (c 0.385 in HOAc).

m/e 146, MH⁺ from CI; $[\alpha]_0^{2^2} + 34.3^\circ$ (c 0.385 in HOAc). (2S,4S)-2-Amino-4-methylhexanoic acid. Refluxing 2S-(+)-1-bromo-2-methylbutane (4.5 g, prepared from 2S-(-)-2-methylbutan-1-ol) with ethyl acetamidomalonate (5.4 g) and NaOEt (0.6 g) in EtOH [3, 4] gave a mixture of (2S,4S) and (2R,4S) ethyl 2-acetamido-2-carbethoxy-4-methylhexanoate as a light yellow oil (5.6 g). Alkaline hydrolysis (10% NaOH, 2 hr), followed by refluxing (2 hr) of the acidified soln yielded a mixture of (2S,4S)

and (2R,4S)-2-acetamido-4-methylhexanoic acid (2.3 g), mp 190-191° from H_2O , $[\alpha]_D^{27} +5.1°$ (c, 2.86 in MeOH). The CI (isobutane) MS showed peaks at m/e 188, MH⁺, 142 (M⁺-COOH), 130 (M⁺-C₄H₉), 99 (M⁺-COOH—MeCO), 57 C₄H₉⁺. An ag. soln of the above acid mixture was adjusted to pH 7 with ammonia and hydrolysed at 38° with powdered hog kidney acylase I (12 mg) [5, 6] overnight. The soln was acidified to pH 5, filtered, passed through a Zeocarb 225 cation exchange resin column in the H⁺ form, and the column washed with H₂O. The aq. eluates from the column were combined and evapd to dryness. Recrystallization of the solid from H₂O gave (2R,4S) 2-acetamido-4-methylhexanoic acid (0.4 g), mp 192°. $[\alpha]_D^{30}$ +22.3° (c, 6.27 in MeOH). Found: C, 58.0; H, 9.0; N, 7.3%; $C_0H_{17}NO_3$ requires: C, 57.7; H, 9.1; N, 7.5%. The amino acid was eluted from the column with 2N ammonia, and the soln concd to ca 200 ml. Cooling of the soln gave the lustrous, colourless crystals of (2S,4S)-2-amino-4-methyl-hexanoic acid (0.5 g), mp 240-244°. Found: C, 57.9; H, 10.4; N, 9.6%; C₇H₁₅NO, requires C, 57.9; H, 10.4; N, 9.6%. $[\alpha]_D + 35.4^{\circ}$ (c, 0.41 in glacial AcOH) and $[\alpha]_D^{22}$ +25.7° (c, 1.0 in 5 N HCl). The CI (isobutane) MS showed peaks at m/e 146 MH⁺, 100 (M⁺-COOH), 74 (M^+ - C_5H_{11}) and 57 $C_4H_9^+$, identical with the acid obtained on hydrogenation of 1 (MS, GLC, TLC, mp, mmp). This is in agreement with data recently obtained by Bernasconi et al. [9] who used α-chymotrypsin for the resolution of the C-2 centre. The 2S configuration of our amino acid was also confirmed by GLC of its N-TFA-L-prolyl methyl ester derivative [10].

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^{*} Part 3 in the series 'Constituents of a New Guinea Boletus'. For Part 2 see ref. [2].